



## Diurnal fluxes of HONO above a crop rotation

Sebastian Laufs<sup>1</sup>, Mathieu Cazaunau<sup>2,3</sup>, Patrick Stella<sup>4,5</sup>, Ralf Kurtenbach<sup>1</sup>, Pierre Cellier<sup>4</sup>, Abdelwahid Mellouki<sup>2</sup>, Benjamin Loubet<sup>4</sup>, and Jörg Kleffmann<sup>1</sup>

<sup>1</sup>Physikalische und Theoretische Chemie, Fakultät 4, Bergische Universität Wuppertal, Gaußstraße 20, 42119 Wuppertal, Germany

<sup>2</sup>ICARE-CNRS, 1 C Av. de la Recherche Scientifique, 45071 Orléans CEDEX 2, France

<sup>3</sup>LISA, UMR7583, CNRS, Universités Paris Est Créteil et Paris Diderot, 94010 Créteil, France

<sup>4</sup>UMR ECOSYS, INRA, AgroParisTech, Université Paris-Saclay, 78850, Thiverval-Grignon, France

<sup>5</sup>UMR SADAPT, AgroParisTech, INRA, Université Paris-Saclay, 75005, Paris, France

Correspondence to: Jörg Kleffmann (kleffman@uni-wuppertal.de)

Received: 18 November 2016 – Discussion started: 9 December 2016

Revised: 19 April 2017 – Accepted: 12 May 2017 – Published: 13 June 2017

**Abstract.** Nitrous acid (HONO) fluxes were measured above an agricultural field site near Paris during different seasons. Above bare soil, different crops were measured using the aerodynamic gradient (AG) method. Two LOPAPs (Long Path Absorption Photometer) were used to determine the HONO gradients between two heights. During daytime mainly positive HONO fluxes were observed, which showed strong correlation with the product of the  $\text{NO}_2$  concentration and the long wavelength UV light intensity, expressed by the photolysis frequency  $J(\text{NO}_2)$ . These results are consistent with HONO formation by photosensitized heterogeneous conversion of  $\text{NO}_2$  on soil surfaces as observed in recent laboratory studies. An additional influence of the soil temperature on the HONO flux can be explained by the temperature-dependent HONO adsorption on the soil surface. A parameterization of the HONO flux at this location with  $\text{NO}_2$  concentration,  $J(\text{NO}_2)$ , soil temperature and humidity fits reasonably well all flux observations at this location.

### 1 Introduction

During the last decades, many field measurement campaigns have reported unusually high nitrous acid (HONO) concentrations during daytime for remote (Zhou et al., 2002; Acker et al., 2006a; Sörgel et al., 2011a; Villena et al., 2011; Oswald et al., 2015; Meusel et al., 2016), semi-urban (Neftel et al., 1996; Staffelbach et al., 1997; Kleffmann et al., 2005;

Su et al., 2008; Li et al., 2012; Yang et al., 2014) and urban regions (Kleffmann et al., 2002, 2003, Ren et al., 2003, 2006; Acker et al., 2006b; Elshorbany et al., 2009, 2010; Hou et al., 2016; Lee et al., 2016). These results stimulated laboratory investigations on potential HONO precursors from which the most frequently discussed mechanisms are (i) the photosensitized reduction of nitrogen dioxide ( $\text{NO}_2$ ) by organic material, e.g. humic acids (George et al., 2005; Stemmler et al., 2006; 2007; Sosedova et al., 2011; Han et al., 2016), (ii) the photolysis of adsorbed nitric acid (Zhou et al., 2003, 2011; Laufs and Kleffmann, 2016), (iii) biogenic production of nitrite in soil (Su et al., 2011; Ostwald et al., 2013; Maljanen et al., 2013; Oswald et al., 2015; Scharko et al., 2015; Weber, 2015) and (iv) release of adsorbed HONO from soil surfaces after deposition of strong acids (VandenBoer et al., 2013, 2014, 2015; Donaldson et al., 2014). Another discussed source, the reaction of excited gaseous  $\text{NO}_2$  with water (Li et al., 2008), is of minor importance as demonstrated by laboratory (Crowley and Carl, 1997; Carr et al., 2009; Amedro et al., 2011) and modelling studies (Sörgel et al., 2011b; Czader et al., 2012). Also the photolysis of nitrophenols or similar compounds (Bejan et al., 2006) is meaningful only in polluted areas, where concentrations of these precursors are high. Finally, the gas-phase HONO source by the reaction of  $\text{HO}_2 \cdot \text{H}_2\text{O}$  complexes with  $\text{NO}_2$ , recently postulated by Li et al. (2014), could not be confirmed by the same group in simulation chamber experiments (Li et al., 2015) and is also in conflict with recent aircraft measurements (Ye et al., 2015).

Several field studies point to an atmospheric daytime HONO source by heterogeneous photosensitized reduction of  $\text{NO}_2$  on organic substrates (Kleffmann, 2007). In these studies, calculated daytime HONO sources, determined from HONO levels exceeding theoretical photostationary state (PSS) values, showed high correlations with the photolysis rate coefficient  $J(\text{NO}_2)$  or the irradiance and  $\text{NO}_2$  concentration (Vogel et al., 2003; Su et al., 2008; Elshorbany et al., 2009; Sörgel et al., 2011b; Villena et al., 2011; Wong et al., 2012; Lee et al., 2016; Crilley et al., 2016). However, concentrations are not only controlled by the local ground surfaces source processes but also depend on the convective mixing in the atmosphere, leading to a potential misinterpretation of the correlation results. In addition, the assumed PSS conditions may also not be fulfilled when HONO and its precursors were measured close to their sources (Lee et al., 2013; Crilley et al., 2016).

In contrast, flux measurements are able to give direct information about ground surface production and loss processes and are potentially a better tool to investigate HONO sources in the lower atmosphere. Nowadays, eddy covariance (EC) is the most commonly applied method to measure fluxes between the surface and the atmosphere. The lack of fast and sensitive EC-HONO measurement systems, however, requires the use of indirect methods like the aerodynamic gradient (AG) method that has been described by a number of authors (Thom et al., 1975; Sutton et al., 1993) or the relaxed eddy accumulation method (REA) that was also recently used for HONO (Ren et al., 2011; Zhou et al., 2011; Zhang et al., 2012). Unfortunately, the available flux observations indicate different HONO precursors. Harrison and Kitto (1994) and Ren et al. (2011), for example, found a relationship of the HONO flux with the  $\text{NO}_2$  concentration and also its product with light intensity, whereas Zhou et al. (2011) observed a correlation of the HONO flux with adsorbed nitric acid and short wavelength radiation. A campaign above a grassland spread with manure (Twigg et al., 2011) found no evidence for an  $\text{NO}_2$ -driven mechanism producing upward HONO fluxes at a local field site, although HONO and  $\text{NO}_2$  concentrations were coupled with one another, which indicated a regional connection. Hence, the origin of the ground surface daytime HONO source is still a topic of controversial discussion.

The present study was part of the German–French (DFG/INSU-CNRS) PHOTONA project (PHOTOlytic sources of Nitrous Acid in the atmosphere) with laboratory and field investigations concerning HONO in the troposphere. In the present work only the field campaigns aimed at elucidating unknown sources of HONO by flux measurements above an agricultural field site are described. The measurements were performed during different seasons of the year and above different types of canopies using the AG method and the LOPAP (LOng Path Absorption Photometer) technique.

## 2 Methods

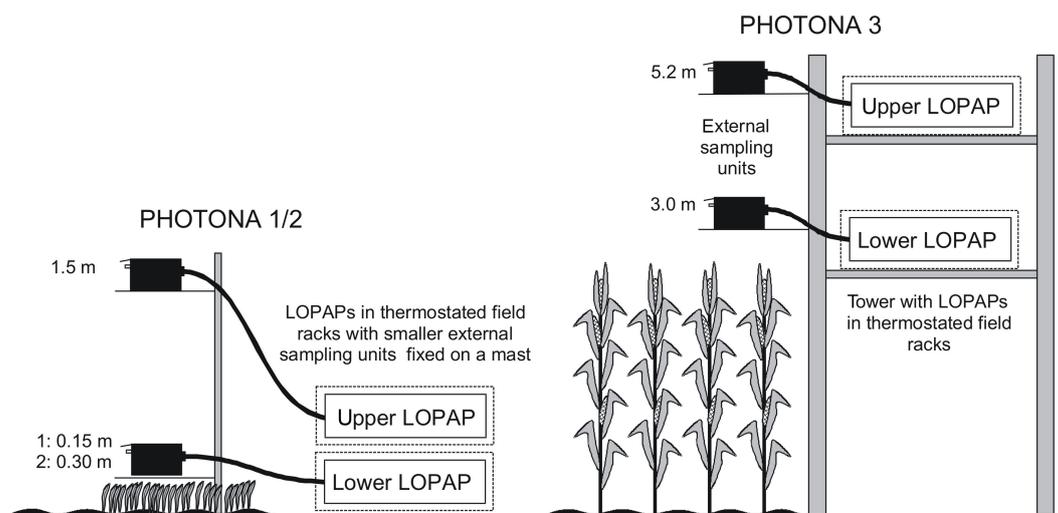
### 2.1 Field site

The measurement site is an agricultural field located in Grignon, around 40 km west from Paris, France (48.9° N, 1.95° E). This site, operated by INRA (Institut National de la Recherche Agronomique) and hosted by AgroParisTech, was part of the EC CarboEurope-IP, NitroEurope-IP, Eclair-IP and INGOS-IP European projects and is part of the ICOS and Fluxnet measurement networks. The site is well documented (Laville et al., 2009; Loubet et al., 2011) as are several experiments on reactive trace gases performed at the site (Bedos et al., 2010; Loubet et al., 2012, 2013; Potier et al., 2015; Personne et al., 2015; Wu et al., 2016; Vuolo et al., 2017). Briefly, measurements were carried out on a 19 ha field with a fetch of 100–400 m depending on wind direction. Roads with substantial traffic surround the site to the south (700 m), east (900 m), north-west (200 m) and south-west (500 m). Other agricultural fields surround the site to the north, south and east. The small village Grignon is located to the west around 700 m away from the measurement site. An animal farm with an average annual production of 210 cattle, 510 sheep and 900 lambs is situated 400 m to the south-south-west. The soil on the field is a silt loam with 31 % clay, 62.5 % silt and 6.5 % sand and was managed with a maize, winter wheat, winter barley and mustard rotation. The field is annually fertilized with nitrogen solution and cattle manure at a rate varying between 100 and 300 kg N ha<sup>-1</sup> yr<sup>-1</sup>, with manure usually applied every 2 to 3 years.

### 2.2 Experimental design

Three field campaigns were performed during the PHOTONA project over a range of crop developments and types: PHOTONA 1 was carried out from 20 to 30 August 2009 over bare soil (for more details see Stella et al., 2012). PHOTONA 2 was a spring campaign from 7 to 19 April 2010 following fertilization with nitrogen solution (composed of 50 % of ammonium nitrate and 50 % of urea, with a nitrogen content of 39 % in dry mass) on 17 March (60 kg N ha<sup>-1</sup>) and 6 April (40 kg N ha<sup>-1</sup>) over a growing (from 0.2 to 0.3 ± 0.05 m height) triticale (hybrid of wheat – *Triticum* – and rye – *Secale*) canopy seeded on 14 October 2009. PHOTONA 3 was a second summer campaign from 16 to 30 August 2011 over a well-developed maize canopy of about 2 m height, seeded on 21 April 2011. The last fertilization of the site before the campaign was on 17 March 2011 with cattle slurry application of 25 m<sup>3</sup> ha<sup>-1</sup>. The slurry composition was 10.5 % dry matter, 28.9 g N kg<sup>-1</sup> of dry matter (of which 12.4 g N kg<sup>-1</sup> was ammonium), corresponding to an amount of nitrogen of 33.4 kg N ha<sup>-1</sup>.

During all campaigns HONO mixing ratios were measured at two heights above the canopy using the LOPAP technique (QUMA Elektronik & Analytik GmbH, Germany), which



**Figure 1.** HONO aerodynamic gradient set-up during the PHOTONA campaigns. Left: PHOTONA 1 (bare soil) and 2 (triticale canopy) with the two external sample units fixed on a mast. Right: scaffold tower during PHOTONA 3 (maize canopy) with the LOPAPs placed at different levels on the tower.

is explained in detail elsewhere (Heland et al., 2001; Kleffmann et al., 2002). The LOPAP instrument allows detection of HONO down to mixing ratios of 1 pptV and the instrument showed excellent agreement with the DOAS (differential optical absorption spectroscopy) technique during inter-comparison studies (Kleffmann et al., 2006). Recently, 15 % interference against  $\text{HNO}_4$  was inferred from laboratory experiments for the LOPAP instrument (Legrand et al., 2014). However, because of the typical high temperatures and low  $\text{NO}_2$  levels during daytime, low  $\text{HNO}_4$  levels (<50 ppt) are expected for the present study, leading to no significant overestimation of the HONO data. In addition, since HONO fluxes are calculated only from the difference of both instruments, potential  $\text{HNO}_4$  interferences are not considered important in the present study. Two LOPAP instruments were placed in thermostated field racks, with the external sampling units fixed at two heights on a mast in the open atmosphere (see Fig. 1). Other trace gases measured during the campaigns were  $\text{NO}_2$  that was detected with the sensitive Luminol technique (LMA-3D or LMA-4, Unisearch Associates Inc., Ontario, Canada), NO (CLD780TR, Ecophys, Switzerland) and  $\text{O}_3$  (FOS, Sextant Technology Ltd, New Zealand), which are based on chemiluminescence techniques. Since the Ecophys  $\text{NO}_x$  instrument is relatively slow and measures NO and  $\text{NO}_x$  consecutively, only the NO channel was used while  $\text{NO}_2$  was detected by the Luminol instrument. Here potential overestimation of  $\text{NO}_2$  by interferences against peroxyacyl nitrates was ignored since their concentrations were unknown.  $\text{NO}_2$ , NO and  $\text{O}_3$  were sampled at a flow rate exceeding  $40 \text{ L min}^{-1}$  with 7 m (9.24 mm internal diameter; i.d.) plus 7 m (3.96 mm i.d.) long PFA (perfluoroalkoxy polymer) sampling lines during PHOTONA 1, 7 m (9.24 mm i.d.) plus 3 m (3.96 mm i.d.) long PFA sam-

pling lines during PHOTONA 2 and 10 m long (9.24 mm i.d.) plus 2 m (3.96 mm i.d.) sampling lines during PHOTONA 3. The residence times in the sampling inlets were estimated to be between 1.6 and 3 s, which were short enough to avoid significant chemical conversions, e.g. by the reaction of  $\text{O}_3$  with NO.

The different canopy heights used during the three campaigns required slight changes in the AG set-up. During PHOTONA 1 and 2, the external sampling units of the LOPAPs were fixed on a small mast at heights of 0.15 and 1.5 m and 0.3 and 1.5 m, respectively (see Fig. 1). Note that the lowest height in PHOTONA 2 was just at the top of the canopy towards the end of the campaign but was always above the displacement height (0.24 m; for definition see Sect. 2.3). The canopy was also quite heterogeneous at that time as shown by the 15 % coefficient of variation of the canopy height and the area around the LOPAP had, on average, a lower canopy height. For PHOTONA 3, a scaffold tower of around 5.5 m height with two levels was installed on the field; on each a LOPAP was mounted (inlet sampling heights 3.0 and 5.2 m). All other trace gases were measured at three heights during PHOTONA 1 (0.2, 0.7, 1.6 m) and 2 (0.4, 0.6, 1.5 m) and one height during PHOTONA 3 (5.0 m), using one instrument for each trace gas connected to Teflon solenoid valves (NRResearch, USA). Measurements were made at 30 s intervals at all three heights (for details see Stella et al., 2012). During all campaigns the sampling inlets were positioned facing away from the field racks towards the prevailing wind direction in order to minimize turbulence disruptions by the racks themselves. For EC measurements a sonic anemometer (R3, Gill Inc., UK) was mounted on a nearby mast at a height of 3.17 m during PHOTONA 1 and 2 and 5.0 m during PHOTONA 3.

Furthermore, meteorological parameters such as wind speed at different heights (cup anemometer, Cimel, FR), wind direction (W200P, Campbell Sci. Inc., USA), relative humidity (RH) and air temperature ( $T_{\text{air}}$ ) (HMP-45, Vaisala, FI) as well as soil parameters like the soil temperature ( $T_{\text{soil}}$ ) (copper–constantan thermocouples) and soil water content at different depths (TDR CS 616, Campbell Sci. Inc., USA) were measured continuously. The photolysis frequency  $J(\text{NO}_2)$  was measured using a filter radiometer (Meteorologie Consult GmbH, Germany) during PHOTONA 1, 2 and 3 and a spectral radiometer (Meteorologie Consult GmbH, Germany) during PHOTONA 3, by which also  $J(\text{O}^1\text{D})$  and  $J(\text{HONO})$  were determined. During PHOTONA 1 and 2  $J(\text{HONO})$  was calculated from measured  $J(\text{NO}_2)$  using the method described by Kraus and Hofzumahaus (1998).

### 2.3 Aerodynamic gradient method

The HONO flux was calculated from the AG method by using a flux–profile relationship based on the Monin–Obukhov (MO) similarity theory that describes the non-dimensional gradient of a scalar  $\chi$  (i.e. the concentration of HONO,  $c(\text{HONO})$ ) as a universal function of the atmospheric stability parameter  $(z-d)/L$  (e.g. Kaimal and Finnigan, 1994).

$$\frac{\kappa \cdot (z-d)}{\chi_*} \cdot \frac{\partial \chi}{\partial z} = \phi_{(z-d)/L} \quad (1)$$

Here,  $\kappa$  is the von Karman constant (0.41),  $\chi$  the measured scalar,  $\chi_*$  the scaling parameter of  $\chi$ ,  $z$  the measurement height,  $d$  the displacement height and  $L$  the Obukhov length. The displacement height accounts for the disturbance of the canopy on the flow and was taken as  $0.7 \cdot h_c$  ( $h_c$  is the height of canopy), a common parameterization in micrometeorology which was validated for this field site by Loubet et al. (2013). During the 1960s and 1970s a lot of effort was spent in the determination of the universal function  $\phi_{(z-d)/L}$  and its primitive  $\Psi_{(z-d)/L}$  (Swinbank, 1964, 1968; Businger et al., 1971). In the actual work, the universal function for heat  $\Psi_{H,(z-d)/L}$  as published by Businger (1966) was integrated with the method of Dyer and Hicks (1970) for the unstable case. For the stable case the universal function for heat as published by Webb (1970) was used (see Supplement).

The flux of a scalar, which is equal to  $u_* \cdot \chi_*$ , can be deduced from Eq. (1), which leads for HONO to (see Supplement for the development of this equation)

$$F_{z_{\text{ref}}} = -\kappa \cdot u_* \cdot \frac{\partial c(\text{HONO})}{\partial [\ln(z-d) - \Psi_{(z-d)/L}]} \quad (2)$$

Here,  $c(\text{HONO})$  is the concentration of HONO.  $F_{z_{\text{ref}}}$  is representative of the flux at the geometric mean height of the concentration measurements,  $z_{\text{ref}}$ , which we hence define as  $z_{\text{ref}} = \sqrt{(z_1-d) \cdot (z_2-d)}$ , where  $z_1$  and  $z_2$  are the measurement heights above the ground. The friction velocity  $u_*$  and the Obukhov length  $L$  were calculated from EC measurements as explained in detail in Loubet et al. (2011).

### 2.4 Data treatment

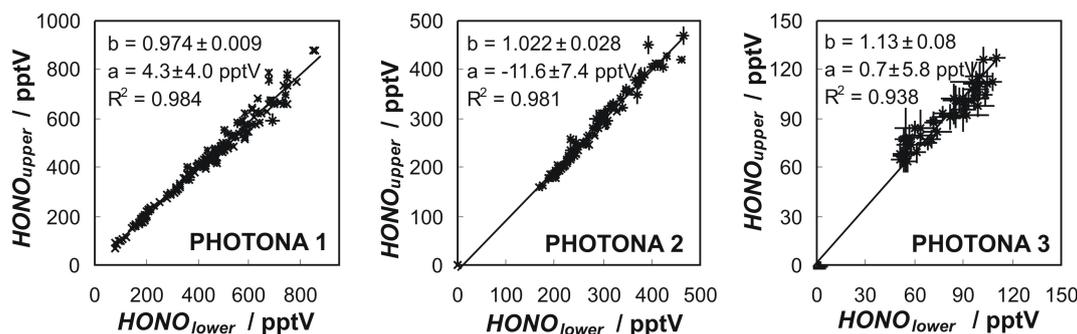
To interpret the flux data for each measurement campaign, first 30 min (PHOTONA 1 and 2) or 5 min (PHOTONA 3) averages were formed from the measurement data including the HONO fluxes. Only 52, 77 and 78 % of the campaign data could be used to determine HONO fluxes during PHOTONA 1, 2 and 3, respectively. For the other periods, data from all instruments were simultaneously not available (calibrations, zeros, intercalibrations, malfunctions, etc.). Secondly, for each campaign a diurnal average day using all this averaged data was calculated by the formation of 1 h means from the whole measurement period. Using this procedure the errors of the individual measurements were reduced by averaging over a large number of values. However, some filtering steps were also applied which removed rain (PHOTONA 1: 24 August 2009; PHOTONA 3: 26 August 2011) and high-emission (PHOTONA 2: 7 April 2010) events from the data. These events led to higher noise in the daily patterns of the HONO flux and were therefore classified as unusual conditions or artefacts that did not represent a common flux profile of the studied agricultural field site. In addition, for PHOTONA 3 the flux data for the first measurement period (16–21 August 2011) were also discarded, caused by low quality of the first intercalibration. From all available flux data, 97, 99 and 57 % were finally used to calculate the average days for PHOTONA 1, 2 and 3, respectively.

Finally, for the correlation analysis of the diurnal average (see Sect. 3.4), weighted orthogonal regression fits (Brauers and Finlayson-Pitts, 1997) of the HONO flux against different variables were applied using the standard error (SE) of the 1 h average for weighting (SE being the standard deviation divided by the square root of  $n$ , the number of data). For the correlations, the lower level data were used for PHOTONA 1 and 2 since they better describe the proposed ground surface source processes. For PHOTONA 3  $\text{NO}_x$  data were available only for one level (5 m), which was used here. To assess the goodness of these fits the merit function  $\chi^2$  and the goodness-of-fit parameter  $Q$  were determined (Brauers and Finlayson-Pitts, 1997). A small  $\chi^2$  and a large  $Q$  indicate a strong linear correlation of the analysed parameters.

### 2.5 Quality of the HONO flux

#### 2.5.1 Estimation of the aerodynamic gradient uncertainty

The following main factors may influence the error of a flux calculation using the AG method. First of all, flux gradient relationships have been studied for quite some time and show good similarities for trace gases such as  $\text{CO}_2$  or sensible and latent heat using the above-described universal functions, but there is always some uncertainty when using an indirect method. Moreover, for HONO the flux similarity has never been compared to other techniques (e.g. the EC method).



**Figure 2.** HONO mixing ratios measured at the same height during PHOTONA 1–3 (from left to right). The solid lines show linear weighted orthogonal regressions (Brauers and Finlayson-Pitts, 1997) between the two instruments. The slope (**b**) and intercept (**a**) are given with their ( $2\sigma$ ) standard deviations.

However, during PHOTONA 1 fluxes of nitric oxide (NO) and ozone ( $O_3$ ) were measured additionally by EC and were in good agreement ( $O_3$ ), or at least comparable (NO) with fluxes calculated by the AG method (Stella et al., 2012). This demonstrated the applicability of the gradient method at the local homogeneous field site.

For the calculation of the uncertainties of the HONO flux by Eq. (2), errors of the gradient ( $\sigma_{\text{gradient}}$ ) and of  $u_*$  ( $\sigma_{u_*}$ ) are of direct importance. During all campaigns, HONO was measured at two heights using two LOPAPs. Hence, the quantification of the gradient strongly depended on the accuracy of these two instruments. The LOPAPs were therefore intercompared several times in the field by placing the external sampling units beside each other and also by using a common PFA inlet line and a T piece between the sampling inlets. In order to estimate the error of both instruments, again weighted orthogonal regressions (Brauers and Finlayson-Pitts, 1997) were applied, using the precision errors of both LOPAPs for weighting (see Fig. 2). The inter-comparisons showed excellent agreement during PHOTONA 1 and 2, with a small intercept and a slope close to 1, demonstrating the capability of the used method to calculate gradients. Not quite so good agreement was obtained for PHOTONA 3, which may partly be explained by the lower HONO levels. To reduce systematic errors in the flux calculation, the lower LOPAP was harmonized using the linear regression fits shown in Fig. 2.

The error of the gradient was then calculated from the precision of the instruments ( $\sigma_{\text{LOPAP}}$ ) and the errors of the slope ( $\Delta b$ ) and the intercept ( $\Delta a$ ) of the regression fit (see Fig. 2), using 95 % confidence intervals ( $2\sigma$ ). The HONO concentration,  $c(\text{HONO})$ , always refers to the higher value of both instruments in order to obtain the maximum deviation.

$$\sigma_{\text{gradient}} = \sqrt{\sigma_{\text{LOPAP}}^2 + \Delta b^2 \cdot c(\text{HONO})^2 + \Delta a^2} \quad (3)$$

The uncertainty of the flux during PHOTONA 1 was finally calculated by error propagation using  $\sigma_{\text{gradient}}$  and  $\sigma_{u_*}$  (for further details of the calculation of  $\sigma_{u_*}$ , see Stella et al., 2012). For PHOTONA 2  $\sigma_{u_*}$  was calculated from 5 min data

of  $u_*$  ( $n = 6$ ). For PHOTONA 3 the uncertainty of the flux was not calculated, as only  $\sigma_{\text{gradient}}$  was available.

### 2.5.2 Influence of the roughness sublayer

The flux gradient similarity is not valid inside the roughness sublayer (RSL), which ranges from the canopy top to around 2 times the canopy height (Cellier and Brunet, 1992). In the present study, the flux divergence caused by the RSL was analysed using the methods of Cellier and Brunet (1992) and Graefe (2004). However, the influence of the RSL during both canopy campaigns was only of minor importance and therefore neglected for further interpretation of the flux data.

### 2.5.3 Dealing with chemical reactions in the gas phase

The aerodynamic gradient method is strictly valid only for non-reactive trace gases. However, in the present study, the photolysis and the production of HONO (e.g. by  $\text{NO} + \text{OH}$ ) in the gas phase below the measurement heights may create artificial fluxes that need to be corrected for.

To check for chemical reactions during turbulent transport the so-called Damköhler number ( $Da$ ) has been used:  $Da = \tau_{\text{trans}}/\tau_{\text{chem}}$ . It compares the chemical reaction timescale ( $\tau_{\text{chem}}$ , see Eq. 4) with the transport timescale ( $\tau_{\text{transp}}$ , see Eq. 5) to identify periods when chemical reaction may generate flux divergence. To calculate the chemical timescale of HONO only its photolysis was taken into account, which is the dominant destruction path of HONO during daytime:

$$\tau_{\text{chem}} = \frac{1}{J(\text{HONO})}. \quad (4)$$

In contrast, loss or production rates by the reactions  $\text{HONO} + \text{OH}$  and  $\text{NO} + \text{OH}$  are typically more than an order of magnitude lower than the HONO photolysis even when considering a typical maximum OH concentration of  $5 \times 10^6 \text{ cm}^{-3}$  during daytime. For the correction of chemistry, the transport timescale depends on (a) the location of the HONO source and (b) the region where HONO chemistry starts to become important. Since photolysis is diminished in

the canopy due to shadowing by the leaves, we only consider the transfer time between the canopy exchange height (defined as  $d + z'_0$ ) and the reference height ( $z_{\text{ref}}$ ). This leads to the following definition of  $\tau_{\text{trans}}$ :

$$\tau_{\text{trans}} = \frac{1}{R_a \cdot (z_{\text{ref}} - d - z_0) + R_b \cdot (z_0 - z'_0)}, \quad (5)$$

where  $R_a$  is the aerodynamic resistance for transfer between  $d + z_0$  ( $z_0$  is the roughness height for momentum) and  $z_{\text{ref}}$ .  $R_b$  is the canopy boundary layer resistance for HONO accounting for the transfer between the roughness height  $d + z_0$  and the canopy exchange height located at  $d + z'_0$  ( $z'_0$  is the roughness height for the scalar).  $R_a$  and  $R_b$  were estimated by

$$R_a = \frac{u_{z_{\text{ref}}}}{(u_*)^2}, \quad (6)$$

$$R_b = \frac{1.45 \cdot Re^{0.24} \cdot Sc^{0.8}}{u_*}. \quad (7)$$

Here  $u_{z_{\text{ref}}}$  is the wind speed at  $z_{\text{ref}}$ ,  $Re$  is the canopy Reynolds number,  $Re = u_* \cdot z_0 / \nu_a$ , and  $Sc$  the Schmidt number,  $Sc = \nu_a / D(\text{HONO})$ , where  $\nu_a$  is the kinematic viscosity of air and  $D(\text{HONO})$  the diffusivity of HONO in air (Garland et al., 1977). During PHOTONA 3, for which the direct measured photolysis rate  $J(\text{HONO})$  was available, the transport time during daytime was typically of the order of a minute and much smaller than the chemical lifetime of HONO of  $\tau_{\text{chem}} \geq 10$  min. Thus, the influence of the photolytic loss to the overall HONO flux was always below 10 % ( $Da < 0.1$ ) and we considered a refinement of the analysis by the stability corrections on  $R_a$  (see Stella et al., 2012) of less importance. As no production terms for HONO were considered for the calculation of the flux divergence, the influence of photolysis gives even only an upper limit for the flux divergence. Similar results were obtained for PHOTONA 1 and 2 using calculated  $J(\text{HONO})$  data. For further analysis, errors by chemical reactions were neglected, which will, however, not significantly influence the interpretation of potential precursors and driving forces of the HONO flux.

### 2.5.4 Footprint area

The field site in Grignon is quite homogenous although with a slight slope and some building and trees around 600 m to the west. To decide if the flux is influenced by surfaces outside this area that may disturb homogeneity, a footprint analysis, as described by Neftel et al. (2008), has been performed using the model ART Footprint Tool version 1.0, which is available from <http://www.agroscope.admin.ch/art-footprint-tool>. The influence of the field site was >92 % (median of all campaigns) and very comparable to other flux measurements at this location. For example, Loubet et al. (2011) stated that up to 93 % of the field was inside the mast footprint (3.17 m height) during summer/spring campaigns.

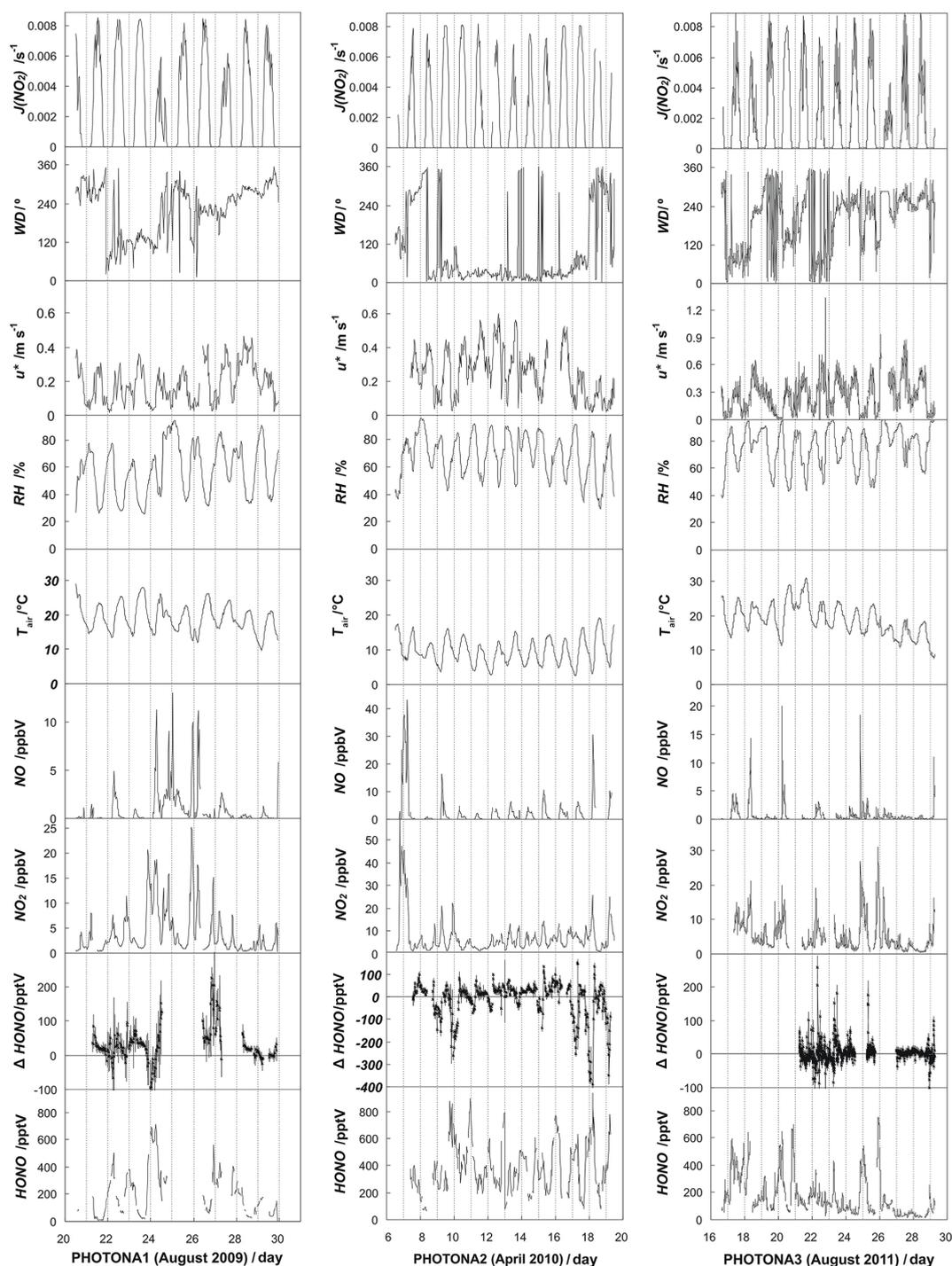
## 3 Results and discussion

### 3.1 General observations

Main standard meteorological measurements and mixing ratios from all campaigns are presented in Fig. 3. During PHOTONA 1, maximum daytime temperatures ranged from 21 to 28 °C and daytime relative humidities were around 30 to 40 %. Minimum night-time temperatures ranged from 10 to 17 °C with relative humidities between 60 and 95 %. Dry conditions generally prevailed with only one moderate rain event on the 24 August (1.2 mm h<sup>-1</sup>). During most of the campaign the wind came from the south-west with only a short period of 2 days dominated by north-easterly winds starting on 22 August. Maximum wind speeds during the day varied from 2 to 8 m s<sup>-1</sup>, with friction velocities up to 0.46 m s<sup>-1</sup>. Minimum HONO mixing ratios during the day varied from 5 to 120 pptV with maximum morning peaks of up to 700 pptV. Minimum daytime mixing ratios of NO<sub>2</sub> were around 1 ppbV with maximum mixing ratios during the morning hours of up to 25 ppbV. Nocturnal mixing ratios of NO<sub>2</sub> varied typically from <1 ppbV in the middle of the night and up to 25 ppbV in the late evening.

Maximum daytime temperatures during PHOTONA 2 varied from 10 to 19 °C with relative humidities in the range 31–65 %. Minimum nocturnal temperatures, reached before early morning, ranged from 2.4 to 7.7 °C with relative humidities between 76 and 93 %. Moderate rain events (up to 1.2 mm h<sup>-1</sup>) occurred during the beginning (8 April) and in the middle (13 April) of the campaign. Maximum wind speeds during the day varied from 3 to 9 m s<sup>-1</sup>, i.e. comparable to PHOTONA 1. However, the canopy generated turbulence, which is expressed in higher friction velocities with maximum values up to 0.6 m s<sup>-1</sup>. HONO mixing ratios during early afternoon varied from 60 to 450 pptV and reached up to 900 pptV during the night. For NO<sub>2</sub> one short emission event with air mass coming from the Paris region occurred during the first day with mixing ratios of up to 60 ppbV. Thereafter, the mixing ratios varied from <20 ppbV during morning hours to around 1 ppbV in the afternoon.

For the first half of the PHOTONA 3 campaign, maximum daytime temperatures reached values up to 31 °C but decreased to around 16 °C during the second half of the campaign. A similar trend was observed for the nocturnal temperatures with minimum values in the range 16 to 22 °C in the beginning and 6 to 12 °C at the end of the campaign. Minimum relative humidities in the afternoon ranged from 40 to 80 % and maximum humidities during early morning were in the range of 80 to 97 %. Light to strong rain events (0.1–8 mm h<sup>-1</sup>) occurred on 21, 22, 26 and 27 August and led to decreases in the HONO mixing ratios, possibly caused by the high effective solubility of HONO in water. Friction velocities reached values up to 0.6 m s<sup>-1</sup> and were comparable to those of PHOTONA 2. HONO mixing ratios reached values up to 600 pptV during the early morning and decreased

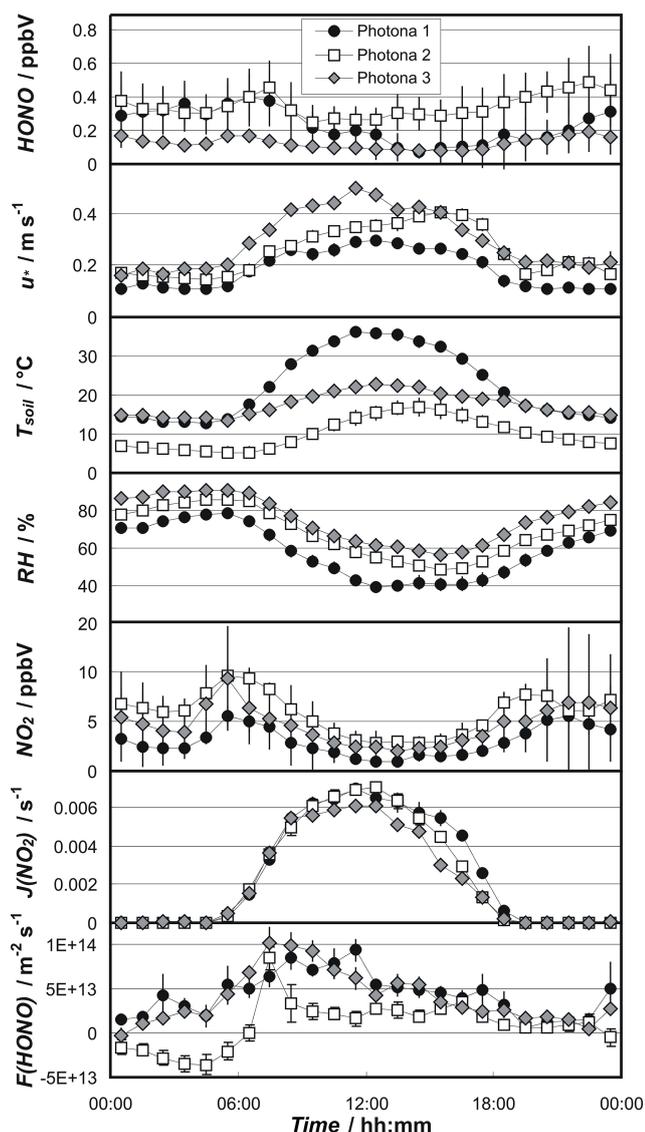


**Figure 3.** Time series of mixing ratios of the main species HONO,  $\text{NO}_2$ , NO and  $\text{O}_3$  (upper level data),  $\Delta \text{HONO}$  (including  $2\sigma$  precision errors), meteorological parameters and  $J(\text{NO}_2)$  during the PHOTONA project (left to right: PHOTONA 1, 2 and 3). WD: wind direction.

to around 20 pptV in the afternoon. Maximum  $\text{NO}_2$  mixing ratios with values up to 30 ppbV were reached in the night or during early morning hours and minimum mixing ratios down to  $< 1$  ppbV were observed in the afternoon.

### 3.2 Diurnal average HONO flux

The HONO flux showed similar profiles in the summer campaigns PHOTONA 1 and 3 but was different in the spring campaign PHOTONA 2 (see Fig. 4). Minimum emissions, or even depositions (PHOTONA 2), occurred at night and



**Figure 4.** Diurnal average data of the HONO flux,  $F(\text{HONO})$ , and its potential precursors and driving factors HONO,  $u_*$ ,  $T_{\text{soil}}$ ,  $c(\text{NO}_2)$  and  $J(\text{NO}_2)$  during the three PHOTONA campaigns.

emissions were observed during daytime with a morning peak at around 08:00 UTC (coordinated universal time). During daytime of the summer campaigns (PHOTONA 1 and 3) continuously decreasing HONO fluxes were observed after the morning peak, whereas during the spring campaign the flux rapidly decreased after a strong morning peak and stayed more or less constant throughout the rest of the day. For PHOTONA 1 and 3 the HONO flux then decreased again to a minimum around midnight or slightly earlier. The magnitudes of the observed daytime fluxes in the range  $0.1$  to  $2.3 \text{ ng N m}^{-2} \text{ s}^{-1}$  ( $0.05$ – $1 \times 10^{14} \text{ molec m}^{-2} \text{ s}^{-1}$ , see Fig. 4) are comparable to measurements of other studies in suburban/rural regions. Ren et al. (2011), for example, found fluxes during daytime with a maximum around

$1.4 \text{ ng N m}^{-2} \text{ s}^{-1}$  on average during CalNex 2010 in California and Zhou et al. (2011) obtained maximum daytime HONO fluxes during noon and early afternoon of around  $2.7 \text{ ng N m}^{-2} \text{ s}^{-1}$  on average on the PROPHET tower in Michigan. The observed morning peak is also in agreement with another study, where the measurements were performed in and above a forest canopy (He et al., 2006) and were explained by dew evaporation.

The range of daytime HONO fluxes measured in this study is also of the order of magnitude of the laboratory derived “optimum HONO emission flux” by biological processes for the soil collected from the Grignon field site, which was  $6.9 \text{ ng N m}^{-2} \text{ s}^{-1}$  (Oswald et al., 2013). In these experiments optimum HONO emissions were derived during drying of the soil surface in a dark chamber by flushing with completely dry air. The cited maximum emission for the Grignon soil was obtained at a soil temperature of  $25^\circ\text{C}$  and at a low soil humidity of around 10 % of the water holding capacity (whc), which corresponds to a gravimetric soil water content of 5.5 % (whc = 54.9 % in gravimetric humidity). Multiplying by the soil density at the surface ( $1.3 \pm 0.5 \text{ kg L}^{-1}$ ) gives the corresponding soil water volume content of 7.1 % much lower than those at the present field site, where soil water contents and soil temperatures at 5 cm depth of  $13.2 \pm 0.4 \%$  and  $22.6 \pm 9.7^\circ\text{C}$  in PHOTONA 1,  $27.1 \pm 2.0 \%$  and  $10.1 \pm 4.2^\circ\text{C}$  in PHOTONA 2 and  $27.7 \pm 1 \%$  and  $18.4 \pm 4.1^\circ\text{C}$  in PHOTONA 3 were observed. According to the soil humidity and temperature response curves reported by Oswald et al. (2013), biological emissions of HONO are expected to be lower than  $5 \text{ ng N m}^{-2} \text{ s}^{-1}$  in PHOTONA 1 and lower than  $0.001 \text{ ng N m}^{-2} \text{ s}^{-1}$  in PHOTONA 2 and PHOTONA 3. Hence we expect the biological source as evaluated by Oswald et al. (2013) to be negligible in PHOTONA 2 and 3, while it could be comparable to the measured HONO flux in PHOTONA 1.

### 3.3 Correlation between fluxes and concentrations of HONO

When plotting the night-time data of the HONO flux against the HONO concentration for PHOTONA 2 (and to lesser extent for PHOTONA 1) a significant positive correlation is observed (PHOTONA 2:  $R^2 = 0.92$ ; PHOTONA 1:  $R^2 = 0.43$ ) with negative HONO fluxes at HONO mixing ratios  $< 0.43$  and  $< 0.13 \text{ ppb}$  for PHOTONA 2 and 1, respectively. This observation indicates a significant impact of HONO deposition on the net HONO flux and is in agreement with the observed negative net HONO fluxes observed in the early morning of PHOTONA 2 (see Fig. 4). In contrast, for the night-time data of PHOTONA 3 and the daytime data of all three campaigns there was no significant correlation between the HONO flux and its concentration. The missing daytime correlation offers support that deposition is of minor importance compared to the more important HONO source terms during daytime.

**Table 1.** Goodness of the weighted orthogonal regressions of hourly average daytime data (06:00 to 20:00 UTC) of  $F(\text{HONO})$  against different variables for the three PHOTONA campaigns. The numbers represent  $\chi^2/Q(R^2)$  values for which lower  $\chi^2$  and higher  $Q$  and  $R^2$  values indicate better correlations (for definition see Brauers and Finlayson-Pitts, 1997). Bold numbers represent the strongest correlations observed for each campaign.

	$J(\text{NO}_2)$	$J(\text{O}^1\text{D})$	$T_{\text{soil}}$	$u_*$	$J(\text{NO}_2) \cdot c(\text{NO}_2)$
PHOTONA 1	27.5/0.004 (0.47)	not measured	$50.2/6 \times 10^{-7}$ (0.22)	23.4/0.016 (0.41)	<b>7.27/0.78</b> <b>(0.79)</b>
PHOTONA 2	12.1/0.28 (0.27)	not measured	9.33/0.50 (0.019)	<b>5.66/0.84</b> <b>(0.37)</b>	12.4/0.26 <b>(0.37)</b>
PHOTONA 3	$53.7/3 \times 10^{-7}$ (0.38)	$79.8/5 \times 10^{-12}$ (0.17)	$121/5 \times 10^{-20}$ (0.03)	$62.7/7 \times 10^{-9}$ (0.20)	<b>3.26/0.994</b> <b>(0.85)</b>

### 3.4 Correlation between fluxes of HONO and potential precursors

As the major aim of the present study was to explain the origin of HONO sources during daytime, the following section concentrates on the flux and its correlations with potential precursors using only data from 06:00 to 20:00 UTC. Campaign averaged HONO fluxes (see Fig. 4) were plotted against different potential precursors and controlling parameters. Correlations of the HONO flux with the product of the photolysis frequency and concentration of  $\text{NO}_2$ ,  $J(\text{NO}_2) \cdot c(\text{NO}_2)$ , were observed for all campaigns (see Table 1). Especially the HONO fluxes during PHOTONA 1 and 3 were well explained by  $\text{NO}_2$  and UV-A light intensity expressed by  $J(\text{NO}_2)$ , which is presented exemplarily for PHOTONA 1 in Fig. 5.

While a correlation between the daytime HONO flux and the product of  $J(\text{NO}_2) \cdot c(\text{NO}_2)$  was observed for all three campaigns, especially during the two summer campaigns PHOTONA 1 and 3, an additional correlation with the friction velocity was observed during the spring campaign PHOTONA 2 (see Table 1). Reasons for the different correlation results and the different diurnal shapes of the HONO fluxes between the two summer and the spring campaigns (see Fig. 4) are still not fully clear. A potential explanation could be the higher influence of HONO deposition during the colder spring campaign (see below) masking the correlation with the main proposed source precursors  $\text{NO}_2$  and radiation. Since deposition fluxes will depend on the turbulent vertical mixing this could explain the higher correlation with the friction velocity. Alternatively, decoupling between the regimes above and below a dense canopy will also depend on the vertical turbulent mixing (Sörgel et al., 2011a) and may have influenced the HONO flux from the soil source region to the measurements heights above the canopy. Finally, stomatal uptake of HONO by the leaves of the triticale canopy, especially during daytime (Schimang et al., 2006), may have caused the lower daytime fluxes during PHOTONA 2 (see Fig. 4) compared to the other campaigns.

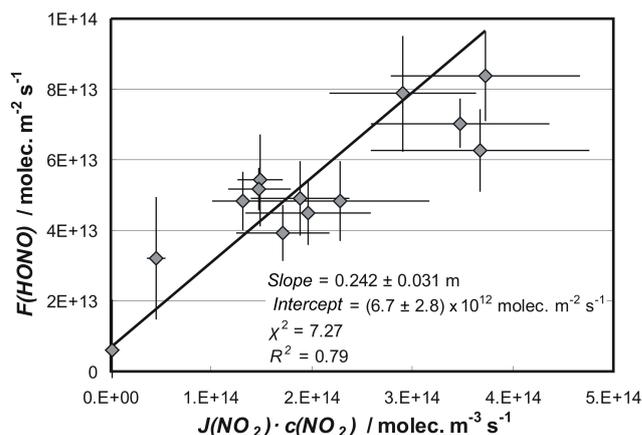
The finding of a light- and  $\text{NO}_2$ -dependent HONO flux is in good agreement with the study of Ren et al. (2011),

where daytime HONO fluxes above an agricultural field also correlated well with the product of  $\text{NO}_2$  concentrations and incident solar radiation during the CalNex 2010 campaign. Also the very weak correlation of the HONO flux with the  $\text{NO}_2$  concentration above a forest canopy at the PROPHET site (Zhou et al., 2011; Zhang et al., 2012) can be attributed to an influence of the canopy. Correlations of HONO with its precursors are expected to become worse when measurements are carried out above high trees as at the PROPHET site, which are able to fully decouple the ground surface from the air above the canopy (Sörgel et al., 2011a; Foken et al., 2012). The results from the present study are qualitatively also in good agreement with former studies in which the daytime source of HONO was quantified using the PSS approach and in which also a strong correlation of the daytime source with radiation and/or  $\text{NO}_2$  was observed (Elshorbany et al., 2009; Sörgel et al., 2011b; Villena et al., 2011; Wong et al., 2012; Lee et al., 2016; Meusel et al., 2016). This observation may imply a mechanism of HONO formation by the reduction of  $\text{NO}_2$  with organic photosensitizer materials like humic acids as proposed in laboratory studies (George et al., 2005; Stemmler et al., 2006, 2007; Han et al., 2016).

Another HONO source, microbiological formation of nitrite in the soil, as proposed by Su et al. (2011) and Oswald et al. (2013), should strongly depend on the soil temperature and the soil surface water content due to the temperature dependence of the solubility of HONO in soil water and/or the adsorption of HONO on the soil surface and the biological activity of the soil. Here, the HONO fluxes are expected to increase with increasing temperature and decreasing humidity. However, except for PHOTONA 2 the correlations of the HONO flux were much weaker with the soil temperature compared to those with  $J(\text{NO}_2)$  and with the product  $J(\text{NO}_2) \cdot c(\text{NO}_2)$  (see Table 1). In addition, the HONO fluxes showed no significant correlation with the soil water content, the relative humidity of the air or its inverse. Also based on the observed diurnal shape of the HONO flux, microbiological formation of nitrite/HONO on the soil surfaces seems to be unlikely, since the highest fluxes would be expected at low soil water content and high temperature, leading to a

maximum of the HONO flux in the early afternoon, when the soil surface is at its driest and warmest due to irradiation from the sun. In contrast, the highest fluxes were observed during the morning in the present study (see Fig. 4). And finally, the expected optimum HONO fluxes were much lower in PHOTONA 3 compared to PHOTONA 1 due mainly to the different soil water contents (see Sect. 3.2), while the measured fluxes were very comparable (see Fig. 4). Thus, although the laboratory derived optimum HONO fluxes were in the same range as those observed in the present field study during PHOTONA 1 (see Sect. 3.2), the different diurnal shapes and seasonal variability of the expected and measured HONO fluxes do not support the microbiological soil mechanism proposed by Su et al. (2011) and Oswald et al. (2013) as a major HONO source at the present field site. This result is in good agreement with another recent field study in which the daytime HONO source could also not be explained by a biological soil source but showed a strong correlation with the radiation (Oswald et al., 2015), similar to that observed in the present study. It should be stressed that in the Oswald et al. (2013) study the experimental conditions were not representative for the present field site. While in these laboratory studies the upper soil surface was flushed by completely dry air, leading to optimum HONO emissions only at very dry conditions, the relative humidity never decreased below 26, 31 and 43% in PHOTONA 1, 2 and 3, respectively. More work is desirable to reconcile HONO field data with incubation experiments as performed by Oswald et al. (2013). Finally, we cannot completely exclude this source here, as we observed a small positive intercept in the correlation plots of the HONO flux against  $J(\text{NO}_2) \cdot c(\text{NO}_2)$  in all campaigns (see Fig. 5 for PHOTONA 1). Since the biological soil source is expected to be light and  $\text{NO}_2$  independent (Su et al., 2011; Oswald et al., 2013) this intercept may reflect the magnitude of this source and/or other light-independent sources. However, the small magnitude of the intercept compared to the observed HONO fluxes, especially for PHOTONA 1 and 3, suggests that light-independent sources are of minor importance during daytime.

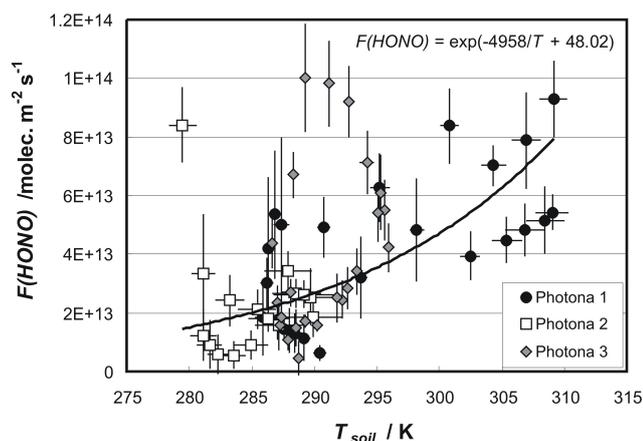
The lack of information about nitrate surface concentrations during the present study does not allow us to directly exclude a  $\text{HNO}_3$  photolysis mechanism as proposed by Zhou et al. (2011), who observed a HONO flux that is positively correlated with leaf surface nitrate loading and light intensity. However, in the present study a better correlation of the HONO flux with  $J(\text{NO}_2)$  (near UV-A) of  $R^2 = 0.38$  than with  $J(O^1D)$  (UV-B) of  $R^2 = 0.17$  was observed for PHOTONA 3 for which a spectroradiometer was used to measure both photolysis frequencies (see Table 1). Since  $\text{HNO}_3$  photolysis is expected to be active mainly under short wavelength UV radiation, while the photosensitized conversion of  $\text{NO}_2$  on humic acid surfaces works well already in the visible and near UV-A (Stemmler et al., 2006; Han et al., 2016), the latter mechanism seems to be a more likely HONO source at the present field site compared to photolysis of ad-



**Figure 5.** Correlation of the diurnal HONO flux (06:00 to 20:00 UTC) with the product  $J(\text{NO}_2) \cdot c(\text{NO}_2)$  during PHOTONA 1 with a weighted orthogonal regression fit (Brauers and Finlayson-Pitts, 1997).

sorbed  $\text{HNO}_3$ . This is confirmed by the high correlation of  $F(\text{HONO})$  with the product  $J(\text{NO}_2) \cdot c(\text{NO}_2)$  of  $R^2 = 0.85$  (see Table 1). In addition, for a potential nitrate photolysis source a maximum of the HONO flux would be expected in the afternoon due to a number of contributing factors: (i) the main  $\text{HNO}_3$  source during daytime is the reaction of  $\text{NO}_2$  with OH, (ii) the typical diurnal profiles of the OH concentration and (iii) the subsequent deposition of gas-phase  $\text{HNO}_3$  onto ground surfaces. In contrast, the campaign averaged HONO fluxes showed asymmetric diurnal profiles with a maximum in the morning, which can be explained by the higher  $\text{NO}_2$  morning levels compared to the afternoon (see Fig. 4). Finally, in a recent laboratory study on the photolysis of adsorbed  $\text{HNO}_3$  only a very low upper-limit photolysis frequency of  $J(\text{HNO}_3 \rightarrow \text{HONO}) = 2.4 \times 10^{-7} \text{ s}^{-1}$  ( $0^\circ$  SZA, 50% RH) was determined (Laufs and Kleffmann, 2016), which is too low to explain any significant HONO formation in the atmosphere.

Another recently discussed mechanism, the acid displacement of HONO by deposition of strong acids (e.g. VandenBoer et al., 2015), also seems to be unlikely for the present field site. This proposed source should maximize in the afternoon because of the daytime formation of the main strong acid  $\text{HNO}_3$  and its subsequent deposition on ground surfaces (see discussion above and see Fig. 4c in VandenBoer et al., 2015). In contrast, for any  $\text{NO}_2$ -dependent photochemical source a maximum HONO flux during morning hours is expected (see Ren et al., 2011, and Fig. 4c in VandenBoer et al., 2015) since the highest  $\text{NO}_2$  concentrations occur in the morning and not in the afternoon (see Fig. 4). Only if the majority of the soil acidity results from night-time dry deposition of  $\text{N}_2\text{O}_5$ , the higher morning fluxes of HONO might be explained by the acid displacement mechanism. Here flux measurements of  $\text{HNO}_3$  and  $\text{N}_2\text{O}_5$  are necessary in the future. However, since we expect a higher contribution



**Figure 6.** Average diurnal HONO fluxes of all individual campaigns as a function of the soil temperature. The black line presents the regression fit using the exponential function  $F(\text{HONO}) = \exp(\Delta_{\text{sol}}H/RT_{\text{soil}} + C)$ , where  $R$  is the universal gas constant ( $8.314 \text{ J mol}^{-1} \text{ K}^{-1}$ ) and  $\Delta_{\text{sol}}H$  is the experimental enthalpy of solvation ( $-41.2 \text{ kJ mol}^{-1}$ ).

of  $\text{HNO}_3$  uptake to the soil acidity, the asymmetric HONO flux profile with higher values in the morning indicates that the acid displacement is of less significance for the present field site (and also for the data shown in Fig. 4c in Vandenberg et al., 2015).

### 3.5 Comparison of all campaigns

In order to find parameters that control the HONO flux in a kind of manner that is not visible using the individual campaign data, we tried to find parameters that affect the HONO flux using the data from all three campaigns. Figure 6 shows HONO fluxes during PHOTONA 1, 2 and 3 as a function of the soil temperature. Although HONO fluxes of the individual campaigns correlated better with  $J(\text{NO}_2) \cdot c(\text{NO}_2)$  (see Table 1), an additional positive correlation of all the data with the soil temperature is obvious. With increasing soil temperature the net HONO flux increases, which may be explained by a temperature-dependent adsorption/solubility process (Su et al., 2011), which becomes more important at lower temperatures compared to the HONO source reactions. In the present study, only net HONO fluxes were quantified, which are controlled by typically smaller negative deposition fluxes and stronger positive formation by heterogeneous processes on the soil surface. When plotting the logarithm of the positive HONO fluxes against the inverse temperature a formal activation enthalpy for HONO formation of  $41.2 \text{ kJ mol}^{-1}$  can be derived (see Fig. 6). Assuming that HONO formation by  $\text{NO}_2$  conversion on the soil surface is controlled by the temperature-dependent HONO solubility in the soil water, this activation enthalpy is in good agreement with the value of the enthalpy of solvation of HONO in water of  $\Delta_{\text{sol}}H = -40.5 \text{ kJ mol}^{-1}$  (Park and Lee, 1988).

The different signs of the two enthalpies are explained by the different reference points describing the same process, for which increasing solubility leads to a decrease in the HONO flux.

In conclusion, positive daytime HONO fluxes are explained in the present study by a  $\text{NO}_2$ - and light-dependent source, i.e. by the photosensitized conversion of  $\text{NO}_2$  on soil surfaces (Stemmler et al., 2006) additionally controlled by the temperature-dependent HONO adsorption on the soil or its solubility in soil water.

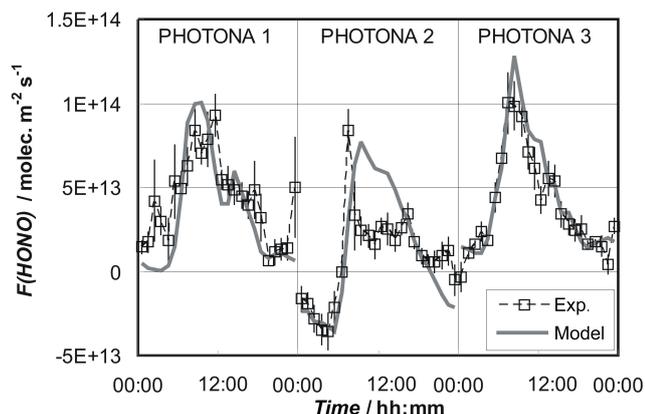
### 3.6 Parameterization of the HONO flux

The above results of the above correlation study were used to set up a simple parameterization that describes the HONO flux for all campaigns. As the strongest correlation was observed for the HONO flux with the product of  $\text{NO}_2$  concentration with light intensity, a proposed photosensitized HONO source (Stemmler et al., 2006) was parameterized by the term  $A \cdot J(\text{NO}_2) \cdot c(\text{NO}_2)$  (see Eq. 8). To also describe the night-time HONO flux, which would have been zero when considering only this light-dependent source, an additional slower dark formation of HONO by heterogeneous  $\text{NO}_2$  conversion on soil surfaces (e.g. Finlayson-Pitts et al., 2003, or Arens et al., 2002) was introduced by using a second source term  $B \cdot c(\text{NO}_2)$ . Because of the observed temperature dependence of the HONO fluxes, both sources were multiplied by a Boltzmann term, for which the negative value of the experimental solvation/adsorption enthalpy of HONO of  $-41.2 \text{ kJ mol}^{-1}$  (see Fig. 6) was used.

The two proposed sources are in agreement with results from several field and laboratory studies (Kleffmann, 2007) but would result in only positive modelled HONO fluxes. However, during PHOTONA 2 also net HONO deposition was observed in the early morning at the low soil temperatures of the spring campaign (see Fig. 4). To describe this net HONO uptake on ground surfaces an additional temperature-dependent HONO deposition term was included, i.e. the product of the HONO concentration measured at the lower sampling height with a temperature-dependent deposition velocity,  $v(\text{HONO})_T$ . Finally, since the magnitude of HONO sources is expected to positively correlate with humidity (Finlayson-Pitts et al., 2003; Stemmler et al., 2006; Han et al., 2016; Su et al., 2011), all variables were optimized for a reference RH of 50% and were scaled linearly with humidity (RH/50%), leading to the final Eq. (8):

$$F(\text{HONO})_{\text{mod}} = [(A \cdot J(\text{NO}_2) \cdot c(\text{NO}_2) + B \cdot c(\text{NO}_2)) \cdot \exp\left(\frac{\Delta_{\text{sol}}H}{R \cdot T_{\text{soil}}}\right) - c(\text{HONO}) \cdot v(\text{HONO})_T] \cdot \frac{\text{RH}}{50\%} \quad (8)$$

The constants  $A$  and  $B$  were adjusted to obtain (i) a slope of one, (ii) an intercept of zero and (iii) a high correlation



**Figure 7.** Diurnally averaged measured HONO fluxes in comparison with modelled values (Eq. 8) during the three PHOTONA campaigns.

between modelled and measured HONO fluxes ( $R^2 = 0.68$ ), resulting in final values for  $A$  and  $B$  of  $2.9 \times 10^6$  m and  $2.0 \times 10^{-3} \text{ m s}^{-1}$ , respectively. When considering also for the Boltzmann and humidity terms, the final value for  $A$  is in good agreement with the average experimental value of  $2.5 \times 10^6$  m determined from the correlation plots of the three campaigns (see Sect. 3.3.2). This indicates that the photolytic HONO source is mainly controlling the net HONO fluxes during daytime. The second term  $B$  multiplied by the Boltzmann and humidity terms can be described as the effective deposition velocity of  $\text{NO}_2$  to form HONO in the dark on ground surfaces. The measured overall deposition velocity of  $\text{NO}_2$  during PHOTONA 1 varied from  $0.002 \text{ m s}^{-1}$  during night time to  $0.0055 \text{ m s}^{-1}$  before noon (calculated from the diurnal average data of the whole campaign; see Stella et al., 2011). Dividing the effective deposition velocity for HONO formation in the dark ( $B$  multiplied by the Boltzmann and humidity terms) by the overall measured deposition velocity of  $\text{NO}_2$ , resulted in campaign averaged ratios in the range 2.0 % (day) to 4.4 % (night), i.e. only 2–4.4 % of the  $\text{NO}_2$  uptake on ground surfaces leads to HONO production by the heterogeneous dark conversion of  $\text{NO}_2$ . This range of values is comparable with night-time observations of Stutz et al. (2002), who calculated a conversion efficiency to form HONO from  $\text{NO}_2$  deposition of  $3 \pm 1$  %.

When comparing the two proposed sources, the dark conversion of  $\text{NO}_2$  contributed only  $\sim 10$  % to the HONO fluxes around noon, while it was the only source during night time by definition. When integrating over the whole day (24 h), the dark conversion contributed 23, 28 and 30 % to the total heterogeneous HONO production, while the photochemical source was 3.3, 2.6 and 2.3 times larger during PHOTONA 1, 2 and 3, respectively. These results are in general agreement with former field studies using the more simple PSS approach in which the photochemical HONO source also dominates daytime production (Kleffmann, 2007, and references

therein), while the dark conversion of  $\text{NO}_2$  controls the night-time build-up of HONO and the OH radical production in the early morning after sunrise (e.g. Alicke et al., 2002).

To describe also the negative HONO fluxes during the PHOTONA 2 spring campaign (see Fig. 4), the temperature-dependent effective HONO deposition velocity ( $\nu(\text{HONO})_T$ , see Eq. 8) was adjusted to values of  $0.02 \text{ m s}^{-1}$  at  $0^\circ\text{C}$  decreasing exponentially to non-significant values at  $40^\circ\text{C}$  ( $\nu(\text{HONO})_T = \exp(23920/T - 91.5)$ ). The higher end deposition velocity used here is in agreement with published upper-limit values in the range  $0.005 \text{ m s}^{-1}$  (Stutz et al., 2002),  $0.017 \text{ m s}^{-1}$  (Harrison and Kitto, 1994; Trebs et al., 2006) and  $0.06 \text{ m s}^{-1}$  (Harrison et al., 1996). Based on this model adjustment, HONO deposition became more significant towards the end of the night, especially during PHOTONA 2, when modelled deposition fluxes were up to 4 times larger compared to the sources. In contrast, during daytime, deposition fluxes were less significant and made up only a few percent at maximum compared to the source reactions, in agreement with the missing correlation of the HONO flux with its concentration during daytime (see Sect. 3.3).

The measured HONO fluxes were well described by Eq. (8) especially during PHOTONA 1 and 3 (see Fig. 7). However, during PHOTONA 2, the campaign with the triticale canopy, the daytime HONO fluxes were overestimated by the model, which may be explained by additional stomatal uptake of HONO by the leaves (Schimang et al., 2006) during transport of HONO from the proposed soil surface source region to the sampling positions above the dense canopy. In addition, the sharp measured morning peak of the HONO flux during PHOTONA 2 is also not well represented by the model. This morning peak may be explained by dew evaporation of accumulated nitrite (formed e.g. by dark reactions of  $\text{NO}_2$  or uptake of HONO) from vegetation surfaces when the temperature increased in the morning, which is in agreement with results from other field studies (Rubio et al., 2002; He et al., 2006).

#### 4 Conclusions

The present study demonstrates the useful application of the aerodynamic gradient method together with the LOPAP technique to measure HONO fluxes over bare soil and canopy surfaces.

Correlation studies of the HONO flux point towards a light-driven HONO source during daytime fed by  $\text{NO}_2$ , which is in line with a photosensitized reaction of  $\text{NO}_2$ , e.g. on humic acid surfaces as observed in laboratory studies. In addition, the comparison of the three campaigns shows an additional influence of the soil temperature on the HONO flux suggesting that adsorption of HONO on the soil surface is of additional importance.

A simple model using two NO<sub>2</sub>-, temperature- and humidity-dependent HONO source terms and a temperature-dependent HONO adsorption was able to reproduce quite satisfactory the measured HONO fluxes, at least for the two PHOTONA summer campaigns. In agreement with known sources of HONO observed in laboratory studies, HONO formation by heterogeneous conversion of NO<sub>2</sub> on ground surfaces is proposed via (i) a slower reaction in the dark and (ii) a faster photosensitized reaction scaling with  $J(\text{NO}_2)$ . The photosensitized source (ii) accounted for ca. 90 % of the daytime HONO formation and was still ca. 3 times stronger compared to (i) when integrated over the whole day in excellent agreement with former field studies using the simpler PSS approach.

*Data availability.* Data used in this publication can be accessed by contacting the corresponding author.

**The Supplement related to this article is available online at <https://doi.org/10.5194/acp-17-6907-2017-supplement>.**

*Competing interests.* The authors declare that they have no conflict of interest.

*Acknowledgements.* The support of this work by the Deutsche Forschungsgemeinschaft (DFG) under contract number (KL 1392/3-1) is gratefully acknowledged. The three experimental campaigns were conducted in the Fluxnet Fr-GRI site supported by INRA. This work was also supported by the French LEFE-CNRS-INSU and R2DS (CNRS, Région Île-de-France) programs and ANR project Vulnoz (ANR-08-VULN-012). The experimental campaigns were also supported by European FP7-NitroEurope (project 017841), FP7-Eclairé (FP7-ENV-2011-282910) and ICOS projects. The authors also gratefully acknowledge Bernard Defransu, Dominique Tristan and Jean-Pierre de Saint-Stéban from the experimental farm of AgroParisTech Grignon for providing access to their fields as well as Michel Burban, Brigitte Durand, Olivier Fanucci, Nicolas Mascher and Jean-Christophe Guedet for their support in the field.

Edited by: J. Roberts

Reviewed by: three anonymous referees

## References

- Acker, K., Möller, D., Wieprecht, W., Meixner, F. X., Bohn, B., Gilge, S., Plass-Dülmer, C., and Berresheim, H.: Strong daytime production of OH from HNO<sub>2</sub> at a rural mountain site, *Geophys. Res. Lett.*, 33, L02809, <https://doi.org/10.1029/2005GL024643>, 2006a.
- Acker, K., Febo, A., Trick, S., Perrino, C., Bruno, P., Wiesen, P., Möller, D., Wiesprecht, W., Auel, R., Guisto, M., Geyer, A.,

- Platt, U., and Allegrini, I.: Nitrous acid in the urban area of Rome, *Atmos. Environ.*, 40, 3123–3133, 2006b.
- Alicke, B., Platt, U., and Stutz, J.: Impact of Nitrous Acid Photolysis on the Total Hydroxy Radical Budget During the Limitation of Oxidant Production/Pianura Padana Produzione di Ozono Study in Milan, *J. Geophys. Res.*, 107, LOP 9-1–LOP 9-17, <https://doi.org/10.1029/2000JD000975>, 2002.
- Amedro, D., Parker, A. E., Schoemaeker, C., and Fittschen, C.: Direct observation of OH radicals after 565 nm multi-photon excitation of NO<sub>2</sub> in the presence of H<sub>2</sub>O, *Chem. Phys. Lett.*, 513, 12–16, 2011.
- Arens, F., Gutzwiller, L., Gäggeler, H. W., and Ammann, M.: The Reaction of NO<sub>2</sub> with Solid Anthrarobin (1,2,10-trihydroxy-anthracene), *Phys. Chem. Chem. Phys.*, 4, 3684–3690, 2002.
- Bedos, C., Rousseau-Djabri, M. F., Loubet, B., Durand, B., Flura, D., Briand, O., and Barriuso, E.: Fungicide volatilization measured and estimated by inverse modelling: the role of vapour pressure and the nature of foliar residue, *Environ. Sci. Technol.*, 44, 2522–2528, 2010.
- Bejan, I., Abd El Aal, Y., Barnes, I., Benter, T., Bohn, B., Wiesen, P., and Kleffmann, J.: The Photolysis of Ortho-Nitrophenols: A new Gas Phase Source of HONO, *Phys. Chem. Chem. Phys.*, 8, 2028–2035, <https://doi.org/10.1039/b516590c>, 2006.
- Brauers, T. and Finlayson-Pitts, B. J.: Analysis of Relative Rate Measurements, *Int. J. Chem. Kinet.*, 29, 665–672, 1997.
- Businger, J. A.: Transfer of momentum and heat in the planetary boundary layer, *Proceedings of the Symposium on Arctic Heat Budget and Atmospheric Circulation (The Rand Corporation)*, 305–332, 1966.
- Businger, J. A., Wyngaard, J. C., Izumi, Y., and Bradley, E. F.: Flux-profile relationships in the atmospheric surface layer, *J. Atmos. Sci.*, 28, 181–189, 1971.
- Carr, S., Heard, D. E., and Blitz, M. A.: Comment on “Atmospheric Hydroxyl Radical Production from Electronically Excited NO<sub>2</sub> and H<sub>2</sub>O”, *Science*, 324, 336b, <https://doi.org/10.1126/science.1166669>, 2009.
- Cellier, P. and Brunet, Y.: Flux gradient relationships above tall plant canopies, *Agric. Forest Meteorol.*, 58, 93–117, 1992.
- Crilley, L. R., Kramer, L., Pope, F. D., Whalley, L. K., Cryer, D. R., Heard, D. E., Lee, J. D., Reed, C., and Bloss, W. J.: On the interpretation of in situ HONO observations via photochemical steady state, *Faraday Discuss.*, 189, 191–212, 2016.
- Crowley, J. N. and Carl, S. A.: OH formation in the photoexcitation of NO<sub>2</sub> beyond the dissociation threshold in the presence of water vapor, *J. Phys. Chem. A*, 101, 4178–4184, 1997.
- Czader, B. H., Rappenglück, B., Percell, P., Byun, D. W., Ngan, F., and Kim, S.: Modeling nitrous acid and its impact on ozone and hydroxyl radical during the Texas Air Quality Study 2006, *Atmos. Chem. Phys.*, 12, 6939–6951, <https://doi.org/10.5194/acp-12-6939-2012>, 2012.
- Donaldson, M. A., Bish, D. L., and Raff, J. D.: Soil surface acidity plays a determining role in the atmosphere-terrestrial exchange of nitrous acid, *P. Natl. Acad. Sci.*, 111, 18472–18477, 2014.
- Dyer, A. and Hicks, B.: Flux-gradient relationships in constant flux layer, *Quart. J. Royal Meteorol. Soc.*, 96, 715–721, 1970.
- Elshorbany, Y. F., Kurtenbach, R., Wiesen, P., Lissi, E., Rubio, M., Villena, G., Gramsch, E., Rickard, A. R., Pilling, M. J., and Kleffmann, J.: Oxidation capacity of the city air of Santiago, Chile,

- Atmos. Chem. Phys., 9, 2257–2273, <https://doi.org/10.5194/acp-9-2257-2009>, 2009.
- Elshorbany, Y. F., Kleffmann, J., Kurtenbach, R., Lissi, E., Rubio, M., Villena, G., Gramsch, E., Rickard, A. R., Pilling, M. J., and Wiesen, P.: Seasonal dependence of the oxidation capacity of the city of Santiago de Chile, Atmos. Environ., 44, 5383–5394, 2010.
- Finlayson-Pitts, B. J., Wingen, L. M., Sumner, A. L., Syomin, D., and Ramazan, K. A.: The Heterogeneous Hydrolysis of NO<sub>2</sub> in Laboratory Systems and in Outdoor and Indoor Atmospheres: An Integrated Mechanism, Phys. Chem. Chem. Phys., 5, 223–242, 2003.
- Foken, T., Meixner, F. X., Falge, E., Zetzsch, C., Serafimovich, A., Bargsten, A., Behrendt, T., Biermann, T., Breuninger, C., Dix, S., Gerken, T., Hunner, M., Lehmann-Pape, L., Hens, K., Jocher, G., Kesselmeier, J., Lüers, J., Mayer, J.-C., Moravek, A., Plake, D., Riederer, M., Rütz, F., Scheibe, M., Siebicke, L., Sörgel, M., Staudt, K., Trebs, I., Tsokankunku, A., Welling, M., Wolff, V., and Zhu, Z.: Coupling processes and exchange of energy and reactive and non-reactive trace gases at a forest site – results of the EGER experiment, Atmos. Chem. Phys., 12, 1923–1950, <https://doi.org/10.5194/acp-12-1923-2012>, 2012.
- Garland, J. A.: Dry deposition of sulfur-dioxide to land and water surfaces, P. Roy. Soc. Lond. Ser. A, 354, 245–268, 1977.
- George, C., Streckowski, R. S., Kleffmann, J., Stemmler, K., and Ammann, M.: Photoenhanced uptake of gaseous NO<sub>2</sub> on solid organic compounds: A photochemical source of HONO?, Faraday Disc., 130, 195–210, 2005.
- Graefe, J.: Roughness layer corrections with emphasis on SVAT model applications, Agr. Forest Meteorol., 124, 237–251, 2004.
- Han, C., Yang, W., Wu, Q., Yang, H., and Xue, X.: Heterogeneous photochemical conversion of NO<sub>2</sub> to HONO on the humic acid surface under simulated sunlight, Environ. Sci. Technol., 50, 5017–5023, 2016.
- Harrison, R. M. and Kitto, A.-M. N.: Evidence for a surface source of atmospheric nitrous acid, Atmos. Environ., 28, 1089–1094, 1994.
- Harrison, R. M., Peak, J. D., and Collins, G. M.: Tropospheric Cycle of Nitrous Acid, J. Geophys. Res., 101, 14429–14439, 1996.
- He, Y., Zhou, X., Hou, J., Gao, H., and Bertman, S. B.: Importance of dew in controlling the air-surface exchange of HONO in rural forested environments, Geophys. Res. Lett., 33, L02813, <https://doi.org/10.1029/2005GL024348>, 2006.
- Heland, J., Kleffmann, J., Kurtenbach, R., and Wiesen, P.: A new instrument to measure gaseous nitrous acid (HONO) in the atmosphere, Environ. Sci. Technol., 35, 3207–3212, 2001.
- Hou, S., Tong, S., Ge, M., and An, J.: Comparison of atmospheric nitrous acid during severe haze and clean periods in Beijing, China, Atmos. Environ., 124, 199–206, 2016.
- Kaimal, J. C. and Finnigan, J. J.: Atmospheric Boundary Layer Flows, Their structure and measurement, Oxford University Press, New York, 1994.
- Kleffmann, J., Heland, J., Kurtenbach, R., Lörzer, J. C., and Wiesen, P.: A new instrument (LOPAP) for the detection of nitrous acid (HONO), Environ. Sci. Poll. Res., special issue 4, 48–54, 2002.
- Kleffmann, J., Kurtenbach, R., Lörzer, J., Wiesen, P., Kalthoff, N., Vogel, B., and Vogel, H.: Measured and simulated vertical profiles of nitrous acid, Part I: Field measurements, Atmos. Environ., 37, 2949–2955, 2003.
- Kleffmann, J., Gavriolaiei, T., Hofzumahaus, A., Holland, F., Koppmann, R., Rupp, L., Schlosser, E., Siese, M., and Wahner, A.: Daytime formation of nitrous acid: A major source of OH-radicals in a forest, Geophys. Res. Lett., 32, L05818, <https://doi.org/10.1029/2005GL022524>, 2005.
- Kleffmann, J., Lörzer, J. C., Wiesen, P., Kern, C., Trick, S., Volkamer, R., Rodenas M., and Wirtz, K.: Intercomparisons of the DOAS and LOPAP techniques for the detection of nitrous acid (HONO), Atmos. Environ., 40, 3640–3652, 2006.
- Kleffmann, J.: Daytime sources of nitrous acid (HONO) in the atmospheric boundary layer, Chem. Phys. Chem., 8, 1137–1144, 2007.
- Kraus, A. and Hofzumahaus, A.: Field measurements of atmospheric photolysis frequencies for O<sub>3</sub>, NO<sub>2</sub>, HCHO, CH<sub>3</sub>CHO, H<sub>2</sub>O<sub>2</sub>, and HONO by UV spectroradiometry, J. Atmos. Chem., 31, 161–180, 1998.
- Laufs, S. and Kleffmann, J.: Investigations on HONO formation from photolysis of adsorbed HNO<sub>3</sub> on quartz glass surfaces, Phys. Chem. Chem. Phys., 18, 9616–9625, 2016.
- Laville, P., Flura, D., Gabrielle, B., Loubet, B., Fanucci, O., Roland, M.-N., and Cellier, P.: Characterization of soil emissions of nitric oxide at field and laboratory scale using high resolution method, Atmos. Environ., 43, 2648–2658, 2009.
- Lee, B. H., Wood, E. C., Herndon, S. C., Lefer, B. L., Luke, W. T., Brune, W. H., Nelson, D. D., Zahniser, M. S., and Munger, J. W.: Urban measurements of atmospheric nitrous acid: A caveat on the interpretation of the HONO photostationary state, J. Geophys. Res.-Atmos., 118, 12274–12281, 2013.
- Lee, J. D., Whalley, L. K., Heard, D. E., Stone, D., Dunmore, R. E., Hamilton, J. F., Young, D. E., Allan, J. D., Laufs, S., and Kleffmann, J.: Detailed budget analysis of HONO in central London reveals a missing daytime source, Atmos. Chem. Phys., 16, 2747–2764, <https://doi.org/10.5194/acp-16-2747-2016>, 2016.
- Legrand, M., Preunkert, S., Frey, M., Bartels-Rausch, Th., Kukui, A., King, M. D., Savarino, J., Kerbrat, M., and Jourdain, B.: Large mixing ratios of atmospheric nitrous acid (HONO) at Concordia (East Antarctic Plateau) in summer: a strong source from surface snow?, Atmos. Chem. Phys., 14, 9963–9976, <https://doi.org/10.5194/acp-14-9963-2014>, 2014.
- Li, S., Matthews, J., and Sinha, A.: Atmospheric hydroxyl radical production from electronically excited NO<sub>2</sub> and H<sub>2</sub>O, Science, 319, 1657–1660, 2008.
- Li, X., Brauers, T., Häsel, R., Bohn, B., Fuchs, H., Hofzumahaus, A., Holland, F., Lou, S., Lu, K. D., Rohrer, F., Hu, M., Zeng, L. M., Zhang, Y. H., Garland, R. M., Su, H., Nowak, A., Wiedensohler, A., Takegawa, N., Shao, M., and Wahner, A.: Exploring the atmospheric chemistry of nitrous acid (HONO) at a rural site in Southern China, Atmos. Chem. Phys., 12, 1497–1513, <https://doi.org/10.5194/acp-12-1497-2012>, 2012.
- Li, X., Rohrer, F., Hofzumahaus, A., Brauers, T., Häsel, R., Bohn, B., Broch, S., Fuchs, H., Gomm, S., Holland, F., Jäger, J., Kaiser, J., Keutsch, F. N., Lohse, I., Lu, K., Tillmann, R., Wegener, R., Wolfe, G. M., Mentel, T. F., Kiendler-Scharr, A., and Wahner, A.: Missing gas-phase source of HONO inferred from Zeppelin measurements in the troposphere, Science, 344, 292–296, 2014.
- Li, X., Rohrer, F., Hofzumahaus, A., Brauers, T., Häsel, R., Bohn, B., Broch, S., Fuchs, H., Gomm, S., Holland, F., Jäger, J., Kaiser, J., Keutsch, F. N., Lohse, I., Lu, K., Tillmann, R., Wegener, R., Wolfe, G. M., Mentel, T. F., Kiendler-Scharr, A., and Wahner, A.:

- Response to Comment on “Missing gas-phase source of HONO inferred from Zeppelin measurements in the troposphere”, *Science*, 348, 1326-e, <https://doi.org/10.1126/science.aaa3777>, 2015.
- Loubet, B., Laville, P., Lehuger, S., Larmanou, E., Fléchar, C., Mascher, N., Genermont, S., Roche, R., Ferrara, R. M., Stella, P., Personne, E., Durand, B., Decuq, C., Flura, D., Masson, S., Fanucci, O., Rampon, J. N., Siemens, J., Kindler, R., Gabrielle, B., Schrupf, M., and Cellier, P.: Carbon, nitrogen and greenhouse gases budgets over a four years crop rotation in northern France, *Plant Soil*, 343, 109–137, 2011.
- Loubet, B., Decuq, C., Personne, E., Massad, R. S., Flechard, C., Fanucci, O., Mascher, N., Gueudet, J.-C., Masson, S., Durand, B., Genermont, S., Fauvel, Y., and Cellier, P.: Investigating the stomatal, cuticular and soil ammonia fluxes over a growing tritic crop under high acidic loads, *Biogeosciences*, 9, 1537–1552, <https://doi.org/10.5194/bg-9-1537-2012>, 2012.
- Loubet, B., Cellier, P., Fléchar, C., Zurfluh, O., Irvine, M., Lamaud, E., Stella, P., Roche, R., Durand, B., Flura, D., Masson, S., Laville, P., Garrigou, D., Personne, E., Chelle, M., and Castell, J.-F.: Investigating discrepancies in heat, CO<sub>2</sub> fluxes and O<sub>3</sub> deposition velocity over maize as measured by the eddy-covariance and the aerodynamic gradient methods, *Agr. Forest Meteorol.*, 169, 35–50, 2013.
- Maljanen, M., Yli-Pirilä, P., Hytönen, J., and Joutsensaari, J.: Acidic northern soils as a source of atmospheric nitrous acid (HONO), *Soil Biol. Biochem.*, 67, 94–97, <https://doi.org/10.1016/j.soilbio.2013.08.013>, 2013.
- Meusel, H., Kuhn, U., Reiffs, A., Mallik, C., Harder, H., Martinez, M., Schuladen, J., Bohn, B., Parchatka, U., Crowley, J. N., Fischer, H., Tomsche, L., Novelli, A., Hoffmann, T., Janssen, R. H. H., Hartogensis, O., Pikridas, M., Vrekoussis, M., Bourtsoukidis, E., Weber, B., Lelieveld, J., Williams, J., Pöschl, U., Cheng, Y., and Su, H.: Daytime formation of nitrous acid at a coastal remote site in Cyprus indicating a common ground source of atmospheric HONO and NO, *Atmos. Chem. Phys.*, 16, 14475–14493, <https://doi.org/10.5194/acp-16-14475-2016>, 2016.
- Nefel, A., Blatter, A., Hesterberg, R., and Staffelbach, T.: Measurements of concentration gradients of HNO<sub>2</sub> and HNO<sub>3</sub> over a semi-natural ecosystem, *Atmos. Environ.*, 30, 3017–3025, 1996.
- Nefel, A., Spirig, C., and Ammann, C.: Application and test of a simple tool for operational footprint evaluations, *Environ. Pollut.*, 152, 644–652, 2008.
- Oswald, R., Behrendt, T., Ermel, M., Wu, D., Su, H., Cheng, Y., Breuninger, C., Moravek, A., Mougou, E., Delon, C., Loubet, B., Pommerening-Röser, A., Sörgel, M., Pöschl, U., Hoffmann, T., Andreae, M. O., Meixner, F. X., and Trebs, I.: HONO emissions from soil bacteria as a major source of atmospheric reactive nitrogen, *Science*, 341, 1233–1235, 2013.
- Oswald, R., Ermel, M., Hens, K., Novelli, A., Ouwersloot, H. G., Paasonen, P., Petäjä, T., Sipilä, M., Keronen, P., Bäck, J., Königstedt, R., Hosaynali Beygi, Z., Fischer, H., Bohn, B., Kubistin, D., Harder, H., Martinez, M., Williams, J., Hoffmann, T., Trebs, I., and Sörgel, M.: A comparison of HONO budgets for two measurement heights at a field station within the boreal forest in Finland, *Atmos. Chem. Phys.*, 15, 799–813, <https://doi.org/10.5194/acp-15-799-2015>, 2015.
- Park, J.-Y. and Lee, Y.-N.: Solubility and decomposition kinetics of nitrous acid in aqueous solution, *J. Phys. Chem.*, 92, 6294–6302, 1988.
- Personne, E., Tardy, F., Genermont, S., Decuq, C., Gueudet, J. C., Mascher, N., Durand, B., Masson, S., Lauransot, M., Flechard, C., Burkhardt, J., and Loubet, B.: Investigating sources and sinks for ammonia exchanges between the atmosphere and a wheat canopy following slurry application with trailing hose, *Agr. Forest Meteorol.*, 207, 11–23, 2015.
- Potier, E., Ogee, J., Jouanguy, J., Lamaud, E., Stella, P., Personne, E., Durand, B., Mascher, N., and Loubet, B.: Multi layer modelling of ozone fluxes on winter wheat reveals large deposition on wet senescing leaves, *Agr. Forest Meteorol.*, 211, 58–71, 2015.
- Ren, X., Harder, H., Martinez, M., Leshner, R. L., Olinger, A., Simpas, J. B., Brune, W. H., Schwab, J. J., Demerjian, K. L., He, Y., Zhou, X., and Gao, H.: OH and HO<sub>2</sub> Chemistry in the urban atmosphere of New York City, *Atmos. Environ.*, 37, 3639–3651, 2003.
- Ren, X., Brune, W. H., Mao, J., Mitchell, M. J., Leshner, R. L., Simpas, J. B., Metcalf, A. R., Schwab, J. J., Cai, C., Li, Y., Demerjian, K. L., Felton, H. D., Boynton, G., Adams, A., Perry, J., He, Y., Zhou, X., and Hou, J.: Behavior of OH and HO<sub>2</sub> in the winter atmosphere in New York City, supplement 2, *Atmos. Environ.*, 40, 252–263, 2006.
- Ren, X., Sanders, J. E., Rajendran, A., Weber, R. J., Goldstein, A. H., Pusede, S. E., Browne, E. C., Min, K.-E., and Cohen, R. C.: A relaxed eddy accumulation system for measuring vertical fluxes of nitrous acid, *Atmos. Meas. Tech.*, 4, 2093–2103, <https://doi.org/10.5194/amt-4-2093-2011>, 2011.
- Rubio, M. A., Lissi, E., and Villena, G.: Nitrite in rain and dew in Santiago City, Chile. Its possible impact on the early morning start of the photochemical smog, *Atmos. Environ.*, 36, 293–297, 2002.
- Scharko, N. K., Schütte, U. M. E., Berke, A. E., Banina, L., Peel, H. R., Donaldson, M. A., Hemmerich, C., White, J. R., and Raff, J. D.: Combined flux chamber and genomic approach links nitrous acid emissions to ammonia oxidizing bacteria and archaea in urban agricultural soil, *Environ. Sci. Technol.*, 49, 13825–13834, 2015.
- Schimang, R., Folkers, A., Kleffmann, J., Kleist, E., Miebach, M., and Wildt, J.: Uptake of Gaseous Nitrous Acid (HONO) by Several Plant Species, *Atmos. Environ.*, 40, 1324–1335, 2006.
- Sörgel, M., Trebs, I., Serafimovich, A., Moravek, A., Held, A., and Zetzsch, C.: Simultaneous HONO measurements in and above a forest canopy: influence of turbulent exchange on mixing ratio differences, *Atmos. Chem. Phys.*, 11, 841–855, <https://doi.org/10.5194/acp-11-841-2011>, 2011a.
- Sörgel, M., Regelin, E., Bozem, H., Diesch, J.-M., Drewnick, F., Fischer, H., Harder, H., Held, A., Hosaynali-Beygi, Z., Martinez, M., and Zetzsch, C.: Quantification of the unknown HONO daytime source and its relation to NO<sub>2</sub>, *Atmos. Chem. Phys.*, 11, 10433–10447, <https://doi.org/10.5194/acp-11-10433-2011>, 2011b.
- Sosedova, Y., Rouvière, A., Bartels-Rausch, T., and Ammann, M.: UVA/Vis induced nitrous acid formation on polyphenolic films exposed to gaseous NO<sub>2</sub>, *Photochem. Photobiol. Sci.*, 10, 1680–1690, 2011.

- Staffelbach, T., Neftel, A., and Horowitz, L. W.: Photochemical oxidant formation over southern Switzerland 2. Model results, *J. Geophys. Res.*, 102, 23363–23373, 1997.
- Stella, P., Loubet, B., Laville, P., Lamaud, E., Cazaunau, M., Laufs, S., Bernard, F., Gosselin, B., Mascher, N., Kurtenbach, R., Mellouki, A., Kleffmann, J., and Cellier, P.: Comparison of methods for the determination of NO-O<sub>3</sub>-NO<sub>2</sub> fluxes and chemical interactions over a bare soil, *Atmos. Meas. Tech. Discuss.*, 4, 5481–5527, <https://doi.org/10.5194/amt-d-4-5481-2011>, 2011.
- Stella, P., Loubet, B., Laville, P., Lamaud, E., Cazaunau, M., Laufs, S., Bernard, F., Gosselin, B., Mascher, N., Kurtenbach, R., Mellouki, A., Kleffmann, J., and Cellier, P.: Comparison of methods for the determination of NO-O<sub>3</sub>-NO<sub>2</sub> fluxes and chemical interactions over a bare soil, *Atmos. Meas. Tech.*, 5, 1241–1257, <https://doi.org/10.5194/amt-5-1241-2012>, 2012.
- Stemmler, K., Ammann, M., Dondors, C., Kleffmann, J., and George, C.: Photosensitized reduction of nitrogen dioxide on humic acid as a source of nitrous acid, *Nature*, 440, 195–198, 2006.
- Stemmler, K., Ndour, M., Elshorbany, Y., Kleffmann, J., D'Anna, B., George, C., Bohn, B., and Ammann, M.: Light induced conversion of nitrogen dioxide into nitrous acid on submicron humic acid aerosol, *Atmos. Chem. Phys.*, 7, 4237–4248, <https://doi.org/10.5194/acp-7-4237-2007>, 2007.
- Stutz, J., Alicke, B., and Neftel, A.: Nitrous acid formation in the urban atmosphere: Gradient measurements of NO<sub>2</sub> and HONO over grass in Milan, Italy, *J. Geophys. Res.*, 107, 8192, <https://doi.org/10.1029/2001JD000390>, 2002.
- Su, H., Cheng, Y. F., Shao, M., Gao, D. F., Yu, Z. Y., Zeng, L. M., Slanina, J., Zhang, Y. H., and Wiedensohler, A.: Nitrous acid (HONO) and its daytime sources at a rural site during the 2004 PRIDE-PRD experiment in China, *J. Geophys. Res.-Atmos.*, 113, D14312, <https://doi.org/10.1029/2007JD009060>, 2008.
- Su, H., Cheng, Y., Oswald, R., Behrendt, T., Trebs, I., Meixner, F. X., Andreae, M. O., Cheng, P., Zhang, Y., and Pöschl, U.: Soil nitrite as a source of atmospheric HONO and OH radicals, *Science*, 333, 1616–1618, 2011.
- Sutton, M., Fowler, D., Hargreaves, K., and Storeton-West, R.: Interactions of NH<sub>3</sub> and SO<sub>2</sub> exchange inferred from simultaneous flux measurements over a wheat canopy, in: *Air Poll. Res. Rep.* 47, edited by: Slanina, J., Angeletti, G., and Beilke, S., 1993.
- Swinbank, W. C.: The Exponential wind profile, *Q. J. Roy. Meteorol. Soc.*, 90, 119–135, 1964.
- Swinbank, W. C.: A comparison between prediction of dimensional analysis for constant-flux layer and observations in unstable conditions, *Q. J. Roy. Meteorol. Soc.*, 94, 460–467, 1968.
- Thom, A. S., Stewart, J. B., Oliver, H. R., and Gash, J. H. C.: Comparison of aerodynamic and energy budget estimates of fluxes over a pine forest, *Q. J. Royal Meteorol. Soc.*, 101, 93–105, 1975.
- Trebs, I., Lara, L. L., Zeri, L. M. M., Gatti, L. V., Artaxo, P., Dlugi, R., Slanina, J., Andreae, M. O., and Meixner, F. X.: Dry and wet deposition of inorganic nitrogen compounds to a tropical pasture site (Rondônia, Brazil), *Atmos. Chem. Phys.*, 6, 447–469, <https://doi.org/10.5194/acp-6-447-2006>, 2006.
- Twigg, M., House, E., Thomas, R., Whitehead, J., Phillips, G., Famulari, D., Fowler, D., Gallagher, M., Cape, J., Sutton, M., and Nemitz, E.: Surface/atmosphere exchange and chemical interactions of reactive nitrogen compounds above a manured grassland, *Agr. Forest Meteorol.*, 151, 1488–1503, 2011.
- VandenBoer, T. C., Brown, S. S., Murphy, J. G., Keene, W. C., Young, C. J., Pszenny, A. A. P., Kim, S., Warneke, C., de Gouw, J. A., Maben, J. R., Wagner, N. L., Riedel, T. P., Thornton, J. A., Wolfe, D. E., Dubé, W. P., Öztürk, F., Brock, C. A., Grossberg, N., Lefer, B., Lerner, B., Middlebrook, A. M., and Roberts, J. M.: Understanding the role of the ground surface in HONO vertical structure: High resolution vertical profiles during NACHTT-11, *J. Geophys. Res.-Atmos.*, 118, 10155–10171, 2013.
- VandenBoer, T. C., Markovic, M. Z., Sanders, J. E., Ren, X., Pusede, S. E., Browne, E. C., Cohen, R. C., Zhang, L., Thomas, J., Brune, W. H., and Murphy, J. G.: Evidence for a nitrous acid (HONO) reservoir at the ground surface in Bakersfield, CA, during CALNex 2010, *J. Geophys. Res.-Atmos.*, 119, 9093–9106, 2014.
- VandenBoer, T. C., Young, C. J., Talukdar, R. K., Markovic, M. Z., Brown, S. S., Roberts, J. M., and Murphy, J. G.: Nocturnal loss and daytime source of nitrous acid through reactive uptake and displacement, *Nat. Geosci.*, 8, 55–60, 2015.
- Villena, G., Wiesen, P., Cantrell, C. A., Flocke, F., Fried, A., Hall, S. R., Hornbrook, R. S., Knapp, D., Kosciuch, E., Mauldin III, R. L., McGrath, J. A., Montzka, D., Richter, D., Ullmann, K., Walega, J., Weibring, P., Weinheimer, A., Staebler, R. M., Liao, J., Huey, L. G., and Kleffmann, J.: Nitrous acid (HONO) during polar spring in Barrow, Alaska: A net source of OH radicals?, *J. Geophys. Res.-Atmos.*, 116, D00R07, <https://doi.org/10.1029/2011JD016643>, 2011.
- Vogel, B., Vogel, H., Kleffmann, J., and Kurtenbach, R.: Measured and Simulated Vertical Profiles of Nitrous Acid – Part II. Model Simulations and Indications for a Photolytic Source, *Atmos. Environ.*, 37, 2957–2966, 2003.
- Vuolo, R. M., Loubet, B., Mascher, N., Gueudet, J.-C., Durand, B., Laville, P., Zurfluh, O., Ciuraru, R., Stella, P., and Trebs, I.: Nitrogen oxides and ozone fluxes from an oilseed-rape management cycle: the influence of cattle slurry application, *Biogeosciences*, 14, 2225–2244, <https://doi.org/10.5194/bg-14-2225-2017>, 2017.
- Webb, E. K.: Profile relationships. Log-linear range, and extension to strong stability, *Q. J. Roy. Meteorol. Soc.*, 96, 67–90, 1970.
- Weber, B., Wu, D., Tamm, A., Ruckteschler, N., Rodriguez-Caballero, E., Steinkamp, J., Meusel, H., Elbert, W., Behrendt, T., Sörgel, M., Cheng, Y., Crutzen, P. J., Su, H., and Pöschel U.: Biological soil crusts accelerate the nitrogen cycle through large NO and HONO emissions in drylands, *P. Natl. Acad. Sci. USA*, 112, 15384–15389, 2015.
- Wong, K. W., Tsai, C., Lefer, B., Haman, C., Grossberg, N., Brune, W. H., Ren, X., Luke, W., and Stutz, J.: Daytime HONO vertical gradients during SHARP 2009 in Houston, TX, *Atmos. Chem. Phys.*, 12, 635–652, <https://doi.org/10.5194/acp-12-635-2012>, 2012.
- Wu, X., Vuichard, N., Ciaia, P., Viovy, N., de Noblet-Ducoudré, N., Wang, X., Magliulo, V., Wattenbach, M., Vitale, L., Di Tommasi, P., Moors, E. J., Jans, W., Elbers, J., Ceschia, E., Tallec, T., Bernhofer, C., Grünwald, T., Moureaux, C., Manise, T., Ligne, A., Cellier, P., Loubet, B., Larmanou, E., and Ripoche, D.: ORCHIDEE-CROP (v0), a new process-based agro-land surface model: model description and evaluation over Europe, *Geosci. Model Dev.*, 9, 857–873, <https://doi.org/10.5194/gmd-9-857-2016>, 2016.
- Yang, Q., Su, H., Cheng, Y., Lu, K., Cheng, P., Gu, J., Guo, S., Hu, M., Zeng, L., Zhu, T., and Zhang, Y.: Daytime HONO forma-

- tion in the suburban area of the megacity Beijing, China, *Science China Chem.*, 57, 1032–1042, 2014.
- Ye, C., Zhou, X., Pu, D., Stutz, J., Festa, J., Spolaor, M., Cantrell, C., Mauldin, R. L., Weinheimer, A., and Haggerty, J.: Comment on “Missing gas-phase source of HONO inferred from Zeppelin measurements in the troposphere”, *Science*, 348, 1326-d, <https://doi.org/10.1126/science.aaa1992>, 2015.
- Zhang, N., Zhou, X., Bertman, S., Tang, D., Alaghmand, M., Shepson, P. B., and Carroll, M. A.: Measurements of ambient HONO concentrations and vertical HONO flux above a northern Michigan forest canopy, *Atmos. Chem. Phys.*, 12, 8285–8296, <https://doi.org/10.5194/acp-12-8285-2012>, 2012.
- Zhou, X., Civerolo, K., Dai, H., Huang, G., Schwab, J., and Demerjian, K. L.: Summertime nitrous acid chemistry in the atmospheric boundary layer at a rural site in New York State, *J. Geophys. Res.*, 107, 4590, <https://doi.org/10.1029/2001JD001539>, 2002.
- Zhou, X., Gao, H., He, Y., Huang, G., Bertman, S. B., Civerolo, K., and Schwab, J.: Nitric acid photolysis on surfaces in low-NO<sub>x</sub> environments: Significant atmospheric implications, *Geophys. Res. Lett.*, 30, 2217, <https://doi.org/10.1029/2003GL018620>, 2003.
- Zhou, X., Zhang, N., TerAvest, M., Tang, D., Hou, J., Bertman, S., Alaghmand, M., Shepson, P. B., Carroll, M. A., Griffith, S., Dusanter, S., and Stevens, P. S.: Nitric acid photolysis on forest canopy surface as a source for tropospheric nitrous acid, *Nat. Geosci.*, 4, 440–443, 2011.